

**THREE PARAMETER EQUATION-1**

Equation Parameters for Water

$$[ A := -1.94 \ B := -4.80 \ C := 6.74 ]$$

Reference Values for Water

$$[ \mu_0 := 1.7916 \text{ MPa s} \ T_0 := 273.15 \text{ K} ]$$

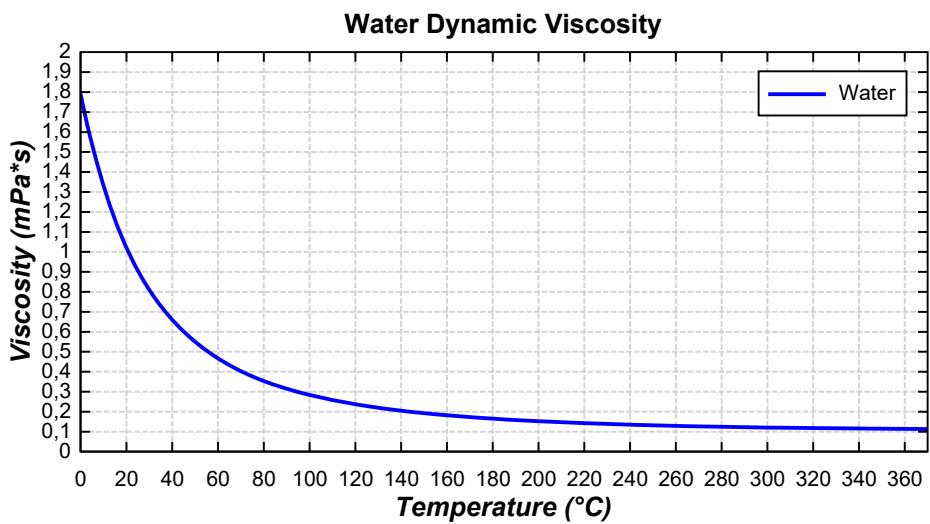
Temperature Table

$$T := [ 0 \dots 370 ] \text{ } ^\circ\text{C}$$

Viscosity Table

$$\mu := \mu_0 \cdot e^{A + B \cdot \left( \frac{T_0}{T} \right) + C \cdot \left( \frac{T_0}{T} \right)^2}$$

Temperature (°C)	Viscosity (mPa · s)
0	1.7916
10	1.3297
20	1.0225
30	0.8107
40	0.6599
50	0.5496
60	0.4669
70	0.4037
80	0.3544
90	0.3153
100	0.2839

**THREE PARAMETER EQUATION-2**

Equation Parameters for Water

$$[ A := -3.7188 \ B := 578.919 \text{ K} \ C := -137.546 \text{ K} ]$$

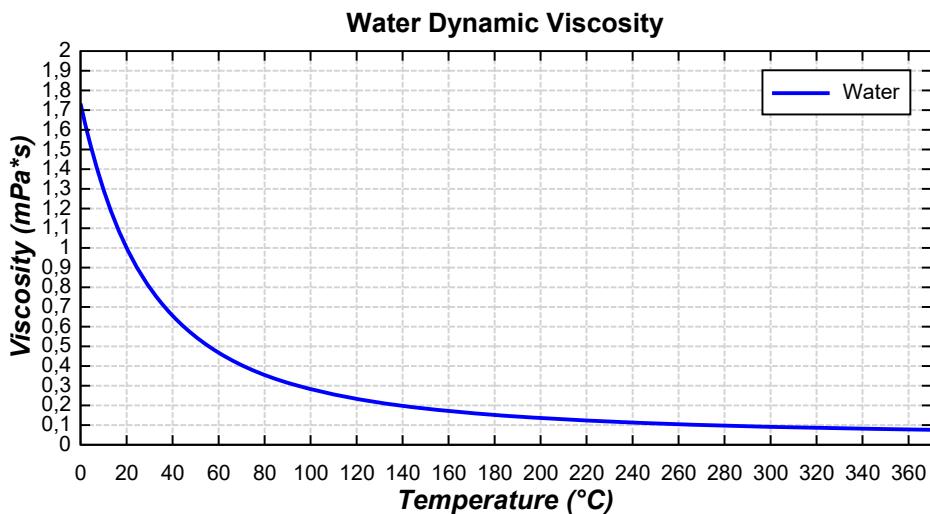
Temperature Table

$$T := [ 0 \dots 370 ] \text{ } ^\circ\text{C}$$

Viscosity Table

$$\eta := e^{A + \frac{B}{C + T}}$$

Temperature (°C)	Viscosity (mPa · s)
0	1.7339
10	1.2933
20	1.0017
30	0.8001
40	0.6557
50	0.549
60	0.4681
70	0.4053
80	0.3557
90	0.3158
100	0.2832



## FOUR PARAMETER EQUATION

Equation Parameters for Water

$$\left[ A := 1.856 \cdot 10^{-11} \quad B := 4209 \quad K \quad C := \frac{0.04527}{K} \quad D := \frac{-3.376 \cdot 10^{-5}}{K^2} \right]$$

Temperature Table

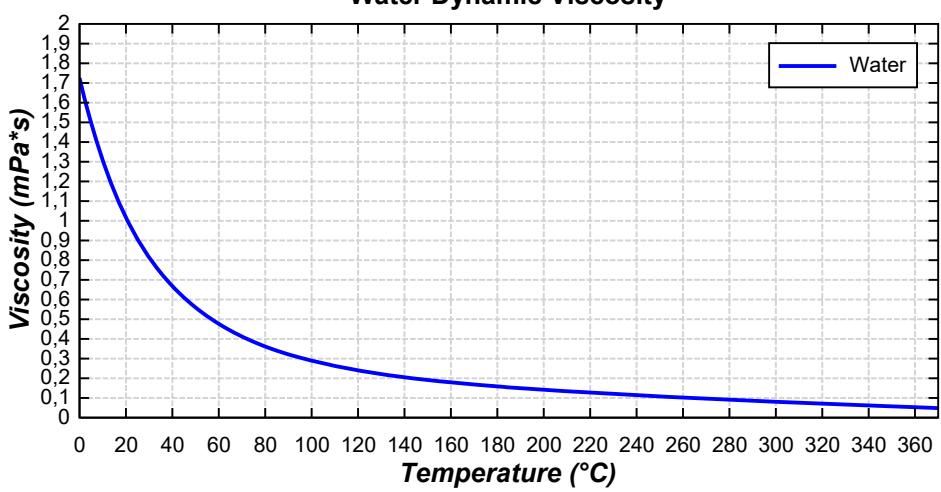
$$T := [0..370] \text{ } ^\circ\text{C}$$

Viscosity Table

$$\eta := A \cdot e^{\frac{B}{T} + C \cdot T + D \cdot T^2}$$

Water Dynamic Viscosity

Temperature (°C)	Viscosity (mPa · s)
0	1.7259
10	1.3053
20	1.0176
30	0.8149
40	0.668
50	0.559
60	0.4765
70	0.4127
80	0.3625
90	0.3223
100	0.2898



## VOGEL-FULSCHER-TAMMANN EQUATION

Equation Parameters for Water

$$[ A := 0.02939 \text{ MPa s} \quad B := 507.88 \text{ K} \quad C := 149.3 \text{ K} ]$$

Temperature Table

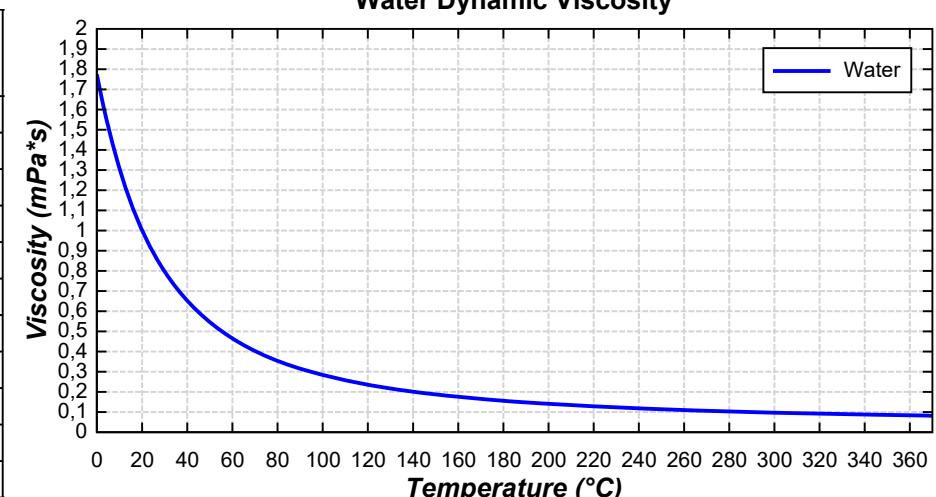
$$T := [0..370] \text{ } ^\circ\text{C}$$

Viscosity Table

$$\mu := A \cdot e^{\frac{B}{(T - C)}}$$

Water Dynamic Viscosity

Temperature (°C)	Viscosity (mPa · s)
0	1.7748
10	1.3064
20	1.0035
30	0.7977
40	0.6522
50	0.5457
60	0.4655
70	0.4037
80	0.355
90	0.316
100	0.2841



**DERIVED EMPIRICAL EQUATION**

Temperature Table

$$T := [0 \dots 370] \text{ } ^\circ\text{C}$$

Formula

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for i ∈ [0 .. 370]
if i < 95
     $\mu_{i+1} := \sqrt{(2.77388442 \cdot i^6 - 1243.59703683 \cdot i^5 + 229813.89243372 \cdot i^4 - 23103721.0686735 \cdot i^3 + 143393546 \cdot i^2 - 45450686) \cdot i^5 + 592.47759433 \cdot i^4 - 315306.5024333 \cdot i^3 + 86868859.3636402 \cdot i^2 - 12933861 \cdot i + 1.7916}$ 
else
     $\mu_{i+1} := \sqrt{(-0.45450686) \cdot i^5 + 592.47759433 \cdot i^4 - 315306.5024333 \cdot i^3 + 86868859.3636402 \cdot i^2 - 12933861 \cdot i + 1.7916}$ 

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Temperature (°C)	Viscosity (mPa · s)
0	1.7916
10	1.3076
20	1.0005
30	0.797
40	0.6539
50	0.5474
60	0.4656
70	0.4024
80	0.3537
90	0.315
100	0.2814

